

## Density Practice Worksheet With Answers

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### Density Practice Worksheet With Answers

Answers 1)  $d = 40.5 \text{ g} / 15.0 \text{ mL}$   $d = 2.7 \text{ g/mL}$  2)  $d = 305.4 \text{ g} / 22.5 \text{ mL}$   $d = 13.57333 \text{ g/mL}$  3)  $d = \text{g} / \text{mL}$   $g = (d) (\text{mL})$   $d = 2.70 \text{ g/mL}$   $d = 13.6 \text{ g/mL}$   $g = (0.789 \text{ g/mL}) (200.0 \text{ mL}) = 158 \text{ g}$  4) (8.4 cm) (5.5 cm) (4.6 cm) = 212.52 cm<sup>3</sup>  $d = 1896 \text{ g} / 212.52 \text{ cm}^3 = 8.9 \text{ g/cm}^3$  Significant figures in the answer are dictated by the length measurements of two sig figs. 5) mass of CCl<sub>4</sub> = 703.55 g - 345.8 g = 357.75 g  $d = 357.75 \text{ g} / 225 \text{ mL} = 1.59 \text{ g/mL}$  6)  $d = 65.14 \text{ g} / 35.4 \text{ mL}$   $d = 1.84 \text{ g/mL}$  7)  $d = \text{g} / \text{mL}$   $g = (d) (\text{mL})$  ...

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$m (\text{silver}) = D \times v$   $m (\text{gold}) = D \times v$ .  $m (\text{silver}) = 10.5 \text{ g/cm}^3 \times 5 \text{ cm}^3$   $m (\text{gold}) = 19.3 \text{ g/cm}^3 \times 5 \text{ cm}^3$ .  $m (\text{silver}) = 52.5 \text{ gm}$   $(\text{gold}) = 96.5 \text{ g}$ . Answer gold. 6. Five mL of ethanol has a mass of 3.9 g, and 5.0 mL of benzene has a mass of 44 g. Which liquid is denser?  $v (\text{ethanol}) = 5 \text{ ml}$   $v (\text{benzene}) = 5 \text{ ml}$ .

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Density Practice Problem Worksheet 1) A block of aluminum occupies a volume of 15.0 mL and weighs 40.5 g. What is its density? 2) Mercury metal is poured into a graduated cylinder that holds exactly 22.5 mL. The mercury used to fill the cylinder weighs 306.0 g. From this information, calculate the density of mercury.

### Density Practice Problem Worksheet - chsd.us

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### Density Math Problems Worksheets - Learny Kids

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Density Practice Problem Worksheet Answers from density calculations worksheet answer key, source:mychaume.com. Informal together with feedback sessions help do away. Adhere about what to edit to the instructions. The estimating worksheet is designed to direct you. There are lots of gantt chart excel templates from the internet.

### Density Calculations Worksheet Answer Key

Find the density of the rock dropped into the graduated cylinder. Finding the density of an irregular shaped object: A graduated cylinder contains 17.5 ml of water. When a metal cube is placed onto the cylinder, its water level rises to 20.3 ml. calculate the density. a. Find the volume of the cube dropped into the graduated cylinder.

### Density Worksheet - Ms. Cyriac's Science Class

Tips for Answering Density Questions. When you're asked to calculate density, make sure your final answer is given in units of mass (such as grams, ounces, pounds, kilograms) per volume (cubic centimeters, liters, gallons, milliliters). You may be asked to give an answer in different units than you're given.

### Density Test Questions with Answers - ThoughtCo

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**Density Calculations 1 Answer - Teacher Worksheets**

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**Practice Density Problems Worksheets - Kiddy Math**

Equation: Density = mass or  $D = \frac{m}{V}$  Problem Statement Formula Define Variables Substitution Answer Sample: What is the density of a billiard ball that has a volume of 100 cm<sup>3</sup> and a mass of 250 g?  $D = \frac{m}{V}$   
 $M = 250 \text{ g}$   $V = 100 \text{ cm}^3$   $D = \frac{250 \text{ g}}{100 \text{ cm}^3} = 2.5 \text{ g/cm}^3$  1. A loaf of bread has a volume of 2270 cm<sup>3</sup> and a mass of 454 g.

**Density Practice: Worksheet #1 - Grizz Physical Science**

This is the basic problem given for understanding DENSITY. Remember that the formula for density is... Density= Mass/ Volume Volume= Mass/ Density Mass= Density × Volume 1. gMass = 40g Volume = 20mL What is the Density? 2 g/ cm<sup>3</sup> 2. Mass = 35g Density = 5 g/ cm<sup>3</sup> What is the Volume? 7 cm<sup>3</sup> 3. Density = 7 g/ mL Volume = 10mL g What is the Mass? 70 g 4.

**Menifee Union School District**

Density Column Worksheet De-6 This one-page student worksheet teaches and reviews the concept of density. The worksheet defines density and gives the mass and volume of four different liquids and two solids. Students calculate the densities of the six substances and complete the density table for mass, volume, and density.

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