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A. 1.20 10[×] 23 B. 3.01 10[×] 23 C. 6.02 10[×] 23 D. 1.20 10[×] 24 2. 1 mol of a hydrocarbon with general formula C_nH_{2n+2} reacts completely with oxygen to produce 4 mol of H₂O. What is the amount of oxygen molecules, in mol, that reacts? A. 4 B. 5 C. 6 D. 7 3. Under which combination of conditions is 1 mol of an ideal gas present? Volume Pressure ...

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The molar mass of a compound is approximately 56 g mol⁻¹. Which formula is possible for this
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2209-6110 - 3 - Turn over M09/4/CHEMI/SPM/ENG/TZ1/XX+ 1. The molar mass of a compound is approximately 56 g mol⁻¹. Which formula is possible for this

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