

## Chapter 4 Atomic Structure Worksheet Answers Pearson

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Chapter 4: Atomic Structure Worksheet . Answer the following questions, circle the best answer. 1) Which of the following are isotopes of each other? a)  $^{14}\text{C}$  and  $^{14}\text{N}$  b)  $^3\text{H}$  and  $^4\text{He}$  c)  $^2\text{H}$  and  $^1\text{H}$  d) none of these . 2) The net charge on an atom that has 13 protons, 12 neutrons, and 10 electrons is . a)  $5+$  b)  $-3$  c)  $3+$  d)  $-9$

### **Name Chapter 4: Atomic Structure Worksheet Answer the ...**

ATOMIC STRUCTURE AND THE PERIODIC TABLE CHAPTER 4 WORKSHEET PART A Given the following isotopes, determine the atomic number, the mass number, the number of protons, electrons and neutrons. Isotope Symbol  $^{131}\text{I}$   $^{51}\text{S}$   $^{238}\text{U}$   $^4\text{He}$   $^{81}\text{W}$   $^{36}\text{Kr}$   $^{37}\text{K}$   $^{37}\text{D}$  Atomic Number Mass Number Protons Electrons Neutrons Isotope Name PARTB.

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Section 4.2 Structure of the Nuclear Atom One change to Dalton's atomic theory is that atoms are divisible into subatomic particles: Electrons, protons, and neutrons are examples of these fundamental particles There are many other types of particles, but we will study these three

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the atom. 3. According to Niels Bohr, what does “n” stand for? 4. Using wavelengths emitted from a hydrogen atom, a student finds that the value of

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On: Defining the Atom, Structure of the Nuclear Atom, Distinguishing Among Atoms. Excuse me while I add in my own comments.

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