

## Atomic Structure Practice Problems Answers

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### Atomic Structure Practice Problems Answers

Practice Problems Atomic Structure and Periodic Properties. 1. How many orbitals are there in a shell of principal quantum number  $n$ ? (Hint: begin with  $n = 1, 2,$  and  $3$  and see if you can recognize the pattern.) Starting with the hint:  $n = 1$  gives  $1$  orbital ( $1s$ )  $n = 2$  gives  $4$  orbitals ( $2s, 2p$ )

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### Atomic Structure Guided Practice Problems Answers

www.njctl.org Chemistry Atomic Structure Answers 1. In the nuclear model of the atom, protons (and neutrons) are housed in a small, dense nucleus. Electrons surround the nucleus in an area of mostly empty space. 2. If electrons are electrically attracted to nucleus and would, therefore, have centripetal acceleration in order to orbit the nucleus.

### Atomic Structure: Chapter Problems - NJCTL

Answers for AP Problems for Chapter 7 - Atomic Structure and Periodicity: 1. A 2. E 3. D 4. C 5. E 6. B 7. B 8. A 9) average = 2.1 a) three points Electron configuration of Na and Mg (1 pt) Any one earns a point: Octet / Noble gas stability comparison of Na and Mg Energy difference explanation between Na and Mg

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Ch.7: Atomic Structure & Periodic Table Trends - Practice Test II Quantum Model & Electronic Structure of Atoms Use the PES spectrum of Phosphorus below to answer questions 1-3. 1. Which peak corresponds to the  $1s$  orbital? (A) 1.06 (B) 1.95 (C)13.5 (D) 18.7 (E) 208 2. Which peak corresponds to the valence  $3p$  orbital?

### Quantum Model & Electronic Structure of Atoms

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